HW01 - Phase Changes/Diagrams & Colligative

Given that you have 14.5 moles of N₂, how many moles of H₂ are theoretically needed to

 $N_2 + 3H_2 \rightarrow 2NH_3$

produce 30.0 moles of NH₃ according to reaction below?

0.75 pts

Properties

Question 1

 \bigcirc 15.0 moles of H₂