

HW05 - Organic Fundamentals

Question 1

5 pts

The inline formula for an ingredient in certain types of nail polish remover is $\text{CH}_3\text{CO}_2\text{CH}_2\text{CH}_3$. What type of compound (classification) is this?

- an aldehyde
- a carboxylic acid
- an ether
- a ketone
- an ester

Question 2

5 pts

What is the name of C_6H_{14} ?

- pentane
- hexene
- propane
- heptane
- hexane

Question 3

5 pts

Which of the following has the greatest molar mass?

- butane
- propane
- nonane
- hexane
- octane
- heptane
- pentane

Question 4

5 pts

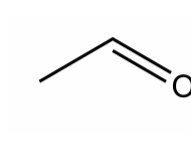
Which of the following has the greatest molar mass?

- methane
- pentane
- pentene
- butane
- butene
- propane

Question 5

5 pts

What is the carbon chain product of the elimination reaction beginning with the reactant shown below?



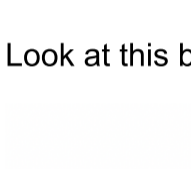
Hint: the eliminated product is HBr. What is left?

- $\text{CHBr}=\text{CHBr}$
- $\text{HC}\equiv\text{CH}$
- $\text{CH}_2\text{Br}-\text{CH}_2\text{Br}$
- $\text{H}_2\text{C}=\text{CH}_2$

Question 6

5 pts

What best describes the functional group on this molecule?

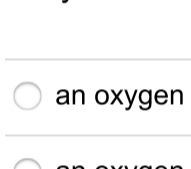


- secondary alcohol
- carboxylic acid
- amine
- ether
- primary alcohol

Question 7

5 pts

What is the functional group on this molecule?

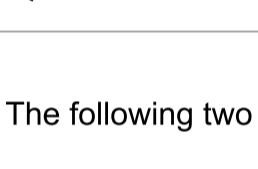


- alcohol
- ketone
- amine
- aldehyde

Question 8

5 pts

What is the functional group on this molecule?

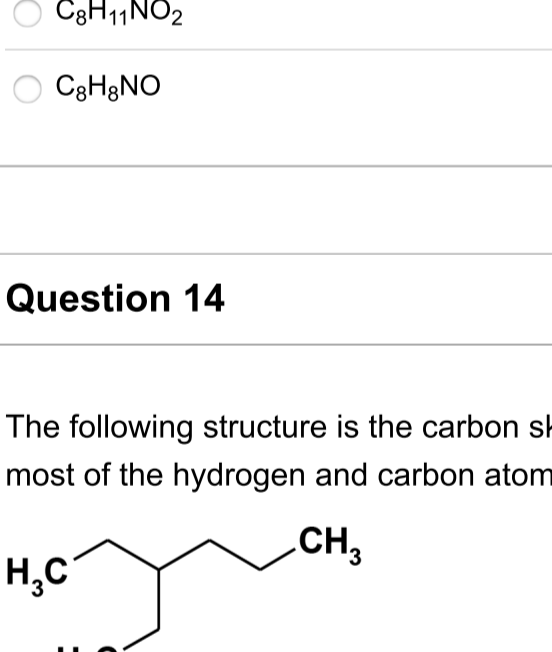


- aldehyde
- alcohol
- ether
- ketone

Question 9

5 pts

Look at this big molecule. What is the functional group on the top right?



- ketone
- carboxylic acid
- aldehyde
- primary amine
- secondary amine

Question 10

5 pts

You are asked to identify the ketone on a large organic molecule. What characteristics are you looking for?

- an oxygen double bonded to a non-terminal carbon
- an oxygen bonded to carbon and hydrogen
- a carbon with a double bond to oxygen and a single bond to an alcohol group
- an oxygen double bonded to a terminal carbon
- a carbon-carbon double bond

Question 11

5 pts

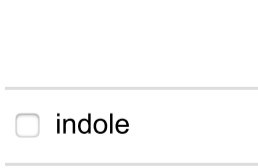
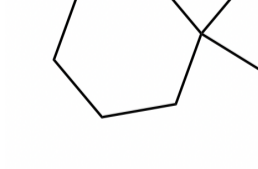
Which of the following is a carboxylic acid functional group?

- R-CHO
- R-COOH
- R-CO
- R-OH
- R-NH₃

Question 12

5 pts

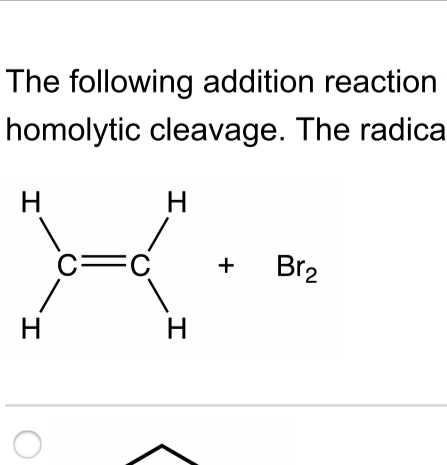
The following two choices both have the formula, $\text{C}_2\text{H}_6\text{O}$. Which is dimethyl ether?



Question 13

6 pts

This is the condensed structural formula for acetaminophen, the active ingredient in the over-the-counter medication Tylenol.



What is the molecular formula of acetaminophen?

- $\text{C}_8\text{H}_9\text{NO}_2$
- $\text{C}_8\text{H}_9\text{NO}_2$
- $\text{C}_9\text{H}_{11}\text{NO}_2$
- $\text{C}_9\text{H}_9\text{NO}$

Question 14

6 pts

The following structure is the carbon skeleton for a structural isomer of octane with most of the hydrogen and carbon atoms omitted.



What is the molecular formula of this structure?

- C_8H_{24}
- C_8H_{16}
- C_8H_{18}
- C_8H_8

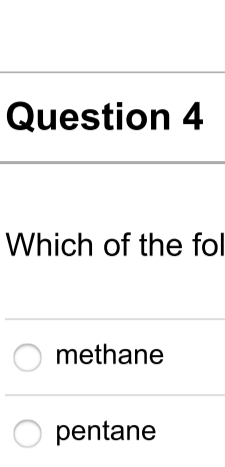
Question 15

6 pts

Consider the following structure:

How many single bonds and double bonds (respectively) are represented by this condensed formula?

Note: the group on the far right can also be read as a phenyl group, similar to what you would see in benzene:

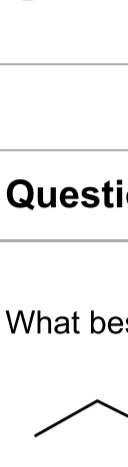


- 15, 4
- 11, 7
- 12, 14
- 12, 4
- 15, 14

Question 16

6 pts

Consider the structural formula of phenol.



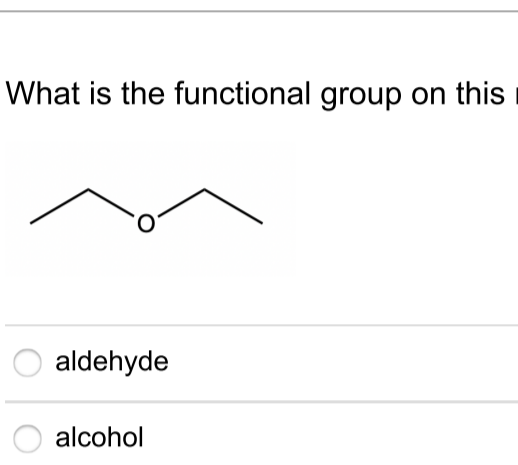
The active ingredient in some oral anesthetics used in sore throat sprays. What is the molar mass of phenol?

- 89 g/mol
- 94 g/mol
- 50 g/mol
- 17 g/mol

Question 17

9 pts

What are three functional groups in the following molecule?

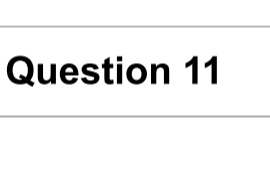
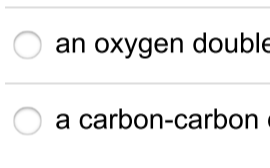
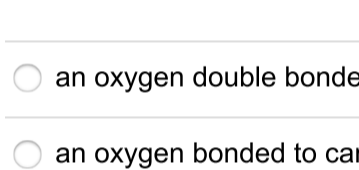
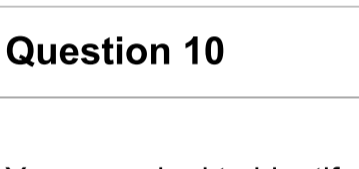
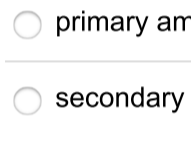


- indole
- ether
- aldehyde
- secondary amine
- alcohol
- primary amine
- carbonyl

Question 18

5 pts

The following addition reaction will proceed when bromine forms two radicals via homolytic cleavage. The radicals then react with ethene to form which product?



Question 19

2 pts

An alkyl halide is placed in solvent and breaks apart to form a carbocation and a halide anion. What type of process is this?

- homolytic cleavage
- heterolytic cleavage