1	Which step of the addition mechanism both increases the length of the polymer chain AND produces a free radical to continue the reaction? perpetuation propagation initiation termination addition
2	Which of the following properly outlines the addition mechanism? Hetereolytic Cleavage - Propagation - Condensation - Termination Initiation - Termination - Propagation Initiation - Propagation - Condensation - Termination Initiation - Propagation - Termination
3	A condensation reaction may occur when which two functional groups are present in the reactants? amine, carboxylic acid aldehyde, ether carboxylic acid, alcohol alcohol, ester ketone, alcohol
4	Which recycle symbol (number) would you most likely find on a large milk container made from the following monomer: $ \begin{pmatrix} H & H \\ C & - C \\ H & H \end{pmatrix}_{n} $
	 1 2 4 6 5
5	LDPE polymers are branched than HDPE, resulting in greater more, strength more, flexibility less, flexibility less, strength
6	6 points Observe the structure below and answer the next two questions.
	A OH OH D
	Which arrow is pointing to a carbonyl carbon? O B O A O C O D
7	On this same structure, which group will be removed in the condensation mechanism? D A B C
8	6 points Fabrics often list their contents in generic terms, rather than proprietary ones. What
	might you find on the care tag of a nylon garment? silk polyamide Kevlar polyester polystyrene
9	4 points Which of the following polymers are made via anaddition reaction mechanism? Polyethylene Nylon Bakelite Polypropylene Polystyrene Polyethylene Terephthalate
10	Polyvinyl Chloride 6 points
	Five of the six "Big 6" plastics are composed of nearly the same repeating monomer, but with differing functional groups substituted into a single position. What is the functional group unique to polypropylene? O amine O halide O methyl O carboxyl O alcohol
11	6 points
	Which of the following functional groups is the distinguishing feature of the monomer used to manufacture styrofoam? O a benzyl group O an amine group O a halide group O a phenyl group O a ester group
12	6 points
	The following three common plastic items are most likely to be composed of which three Big 6 plastics? (identify the plastics by their recycling number) • Disposable coffee cup • Plumbing pipe • Carbonated drink bottle O 2, 4, 6
	5, 2, 33, 1, 46, 3, 1
13	6 points The bakelite polymer consists of phenol and formaldehyde. In the real world, why does this polymer not look as organized as it does in two dimensions? O the carbon-carbon bonds in the phenol groups can rotate and branch in different directions
	 the phenol groups are flat the methylene links are flat and rigid the methylene links can rotate and branch in different directions
14	6 points Which of the following is/are made from amino acid monomers?
	fats wool starch
	cellulose silk biological proteins
15	6 points Which of the following can be glucose polymers?
	proteins flax silk
	cotton carbohydrates wool
16	6 points There are many different types of proteins. What makes a protein unique? O the fact that all amino acids have the same functional groups O the identity of the R-side chain on the amino acid monomers that make up the polymer O the various sugar monomers that make up the protein chain

the carboxylic acid and amine functional groups on the amino acid monomers

Consider the biological polymer of DNA. There are two monomer units (a copolymer)

that make up the backbone chain - what is the repeat unit here?

phosphate + glucose

ester + deoxyribose

peptide link + ribose

phosphate + deoxyribose

phosphate + deoxyfructose

17 6 points