# HW05 - Organic Fundamentals

These questions will get you started on some organic chemistry.

### 1 5 points The inline formula for an ingredient in certain types of nail polish remover is $CH_3CO_2CH_2CH_3$ . What type of compound (classification) is this? $\bigcirc$ an ether $\bigcirc$ an ester $\bigcirc$ an aldehyde $\bigcirc$ a ketone Ο a carboxylic acid 2 5 points What is the name of $C_6H_{14}$ ? Ο hexane $\bigcirc$ propane $\bigcirc$ heptane $\bigcirc$ hexene $\bigcirc$ pentane 5 points 3 Which of the following has the greatest molar mass? $\bigcirc$ heptane Ο pentane Ο propane Ο butane $\bigcirc$ hexane $\bigcirc$ nonane $\bigcirc$ octane 5 points 4 Which of the following has the greatest molar mass? ( ) propane $\bigcirc$ butene

- Ο pentene
- $\bigcirc$ butane
- $\bigcirc$ methane
- Ο pentane

5 5 points

> What is the carbon chain product of the elimination reaction beginning with the reactant shown below?

Br

Hint: the eliminated product is HBr. What is left?

- Ο HC≡CH
- Ο CHBr=CHBr
- $\bigcirc$  $H_2C=CH_2$
- CH<sub>2</sub>Br-CH<sub>2</sub>Br  $\bigcirc$

5 points 6

What best describes the functional group on this molecule?	
ОН	
O primary alcohol	
O amine	
Secondary alcohol	
O ether	
O carboxylic acid	
7 5 points	
What is the functional group on this molecule?	
~ `0	
O ketone	
O amine	
O alcohol	
O aldehyde	
8 5 points	
What is the functional group on this molecule?	
O aldehyde	
O ketone	
O ether	
O alcohol	
9 5 points	
Look at this big molecule. What is the functional group on the top right?	
NH <sub>2</sub>	
NH2	
NH2	

- Ο primary amine Ο ketone
- Ο secondary amine
- $\bigcirc$ aldehyde
- Ο carboxylic acid

### 10 5 points

You are asked to identify the ketone on a large organic molecule. What characteristics are you looking for?

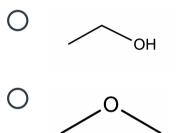
- $\bigcirc$ a carbon-carbon double bond
- $\bigcirc$ an oxygen double bonded to a non-terminal carbon
- $\bigcirc$ an oxygen bonded to carbon and hydrogen
- $\bigcirc$ a carbon with a double bond to oxygen and a single bond to an alcohol group
- Ο an oxygen double bonded to a terminal carbon

## 11 5 points

Which of the following is a carboxylic acid functional group?

- **R-CHO** ()
  - **R-COOH**
- О
- R-OH
- R-CO
- R-NH<sub>3</sub>
- 12 5 points

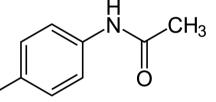
The following two choices both have the formula,  $C_2H_6O$ . Which is dimethyl ether?



13 6 points

HO

This is the condensed structural formula for acetaminophen, the active ingredient in the over-the-counter medication Tylenol.



What is the molecular formula of acetaminophen?

- Ο C<sub>8</sub>H<sub>8</sub>NO
- O C<sub>8</sub>H<sub>11</sub>NO<sub>2</sub>
- $\bigcirc$  C<sub>8</sub>H<sub>9</sub>NO<sub>2</sub>
- $\bigcirc$  C<sub>8</sub>H<sub>5</sub>NO<sub>2</sub>

### 14 6 points

The following structure is the carbon skeleton for a structural isomer of octane with most of the hydrogen and carbon atoms omitted.  $>CH_3$ 

What is the molecular formula of this structure?

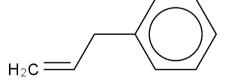
Ο C<sub>8</sub>H<sub>18</sub>  $C_8H_{16}$  $\bigcirc$  $\bigcirc$  $C_8H_{24}$ 

 $C_8H_8$  $\bigcirc$ 

15

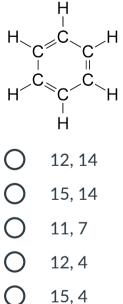
6 points

Consider the following structure:



How many single bonds and double bonds (respectively) are represented by this condensed formula?

Note: the group on the far right can also be read as a phenyl group, similar to what you would see in benzene:



16 6 points

Consider the structural formula of phenol.



OH

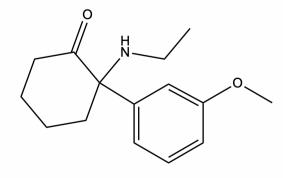
The active ingredient in some oral anesthetics used in sore throat sprays. What is the molar mass of phenol?

- $\bigcirc$ 89 g/mol ()50 g/mol 17 g/mol
- ()94 g/mol

17

9 points

What are three functional groups in the following molecule?

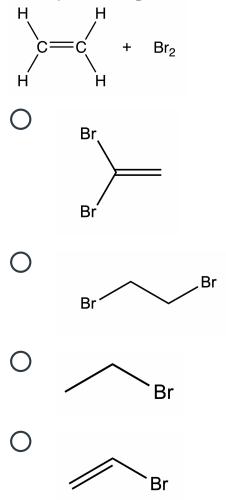


indole

- aldehyde
- carbonyl
- alcohol
- ether
- secondary amine
- primary amine

### 5 points 18

The following addition reaction will proceed when bromine forms two radicals via homolytic cleavage. The radicals then react with ethene to form which product?



2 points 19

> An alkyl halide is placed in solvent and breaks apart to form a carbocation and a halide anion. What type of process is this?

( ) ()

heterolytic cleavage

homolytic cleavage