Nuclear Chemistry

There was a nuclear emergency in Japan on March 11, 2011. The emergency was brought on because the flow of cooling water in nuclear reactors was interrupted. Cooling water interruptions were the cause of the following incidents as well:

- (A) There have been no other situations similar to the Japan situation
- (B) Cooling water was the initial problem at Chernobyl
- (C) Cooling water was the initial problem at 3 Mile Island
- (D) Cooling water was the problem at both Chernobyl and 3 Mile Island

Look at now: Earthquake and Tsunami:

 http://video.nytimes.com/video/2011/03/11/ world/asia/100000000718996/tc-031111.html

This first video didn't play in class, you should have a look at it though. - dr mccord

Look at later: Vlog Brother Explanation:

http://www.youtube.com/watch?v=rBvUtY0PfB8

Did a nuclear explosion occur at the Fukushima plant?

- A) Yes, but only at 4 of the 6 reactors.
- B) No, but experts expect a nuclear explosion to occur if the situation isn't brought under control soon
- C) Yes, they all of the reactors have had nuclear explosions to varying degrees
- D) No, there have been no nuclear explosions, nor do experts believe that a nuclear explosion will occur at the facility

• Status of the 6 reactors:

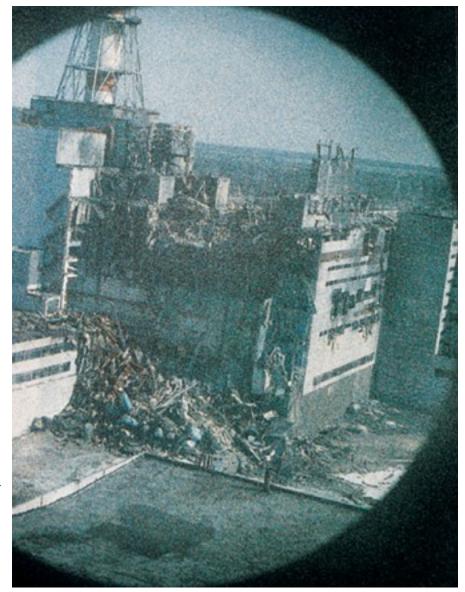
http://www.nytimes.com/interactive/
2011/03/16/world/asia/reactors-status.html

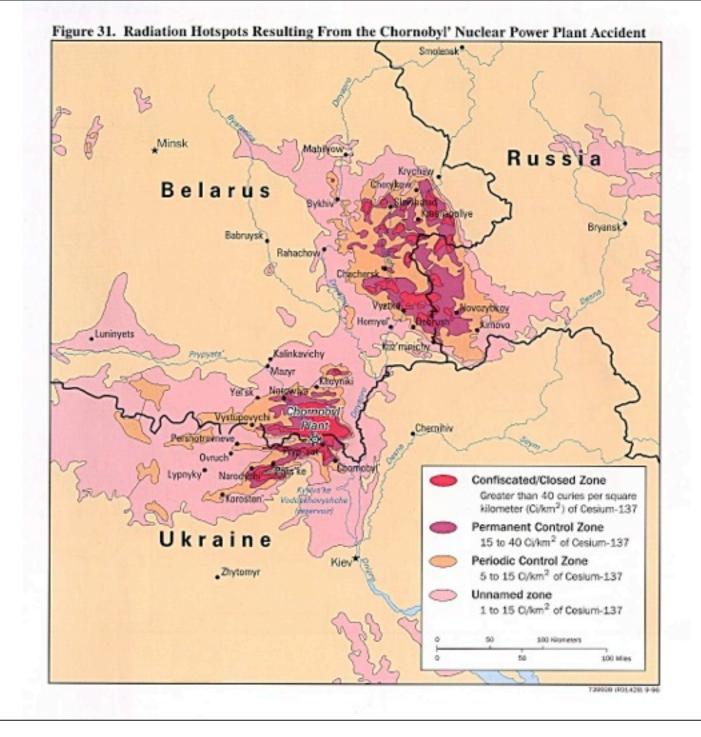
Chernobyl-What Happened: April 26, 1986

Operator error – cooling water mistake

Explosion

9 tons of nuclear material blown into sky 100 times normal background radiation



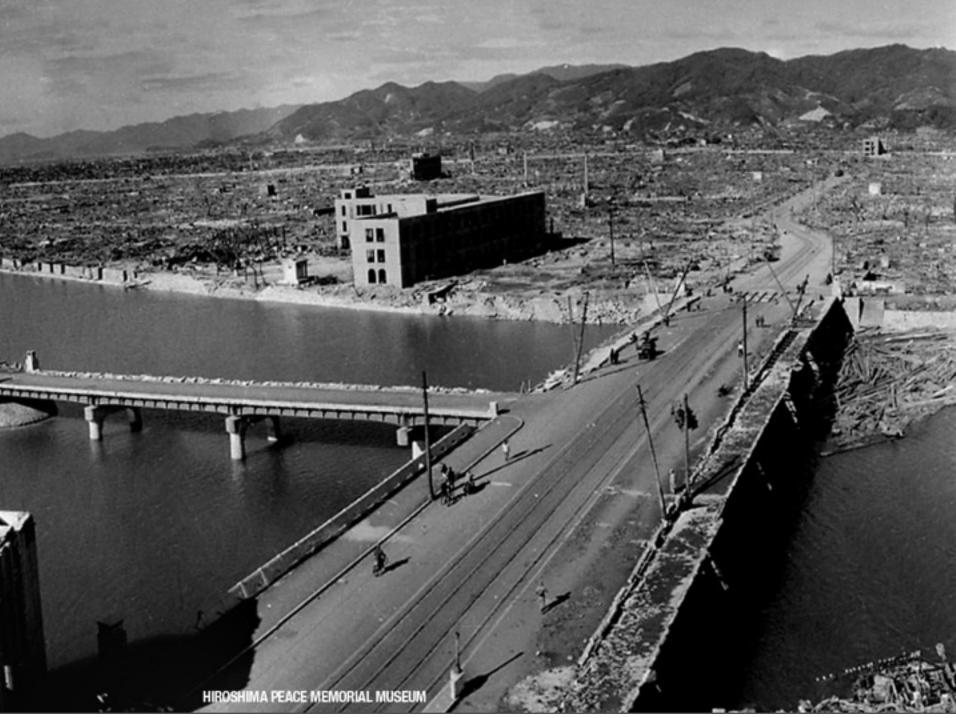


If a nuclear explosion would have happened at the power plant this is what it would have looked like:

http://www.globalsecurity.org/wmd/ops/ hiroshima01.htm



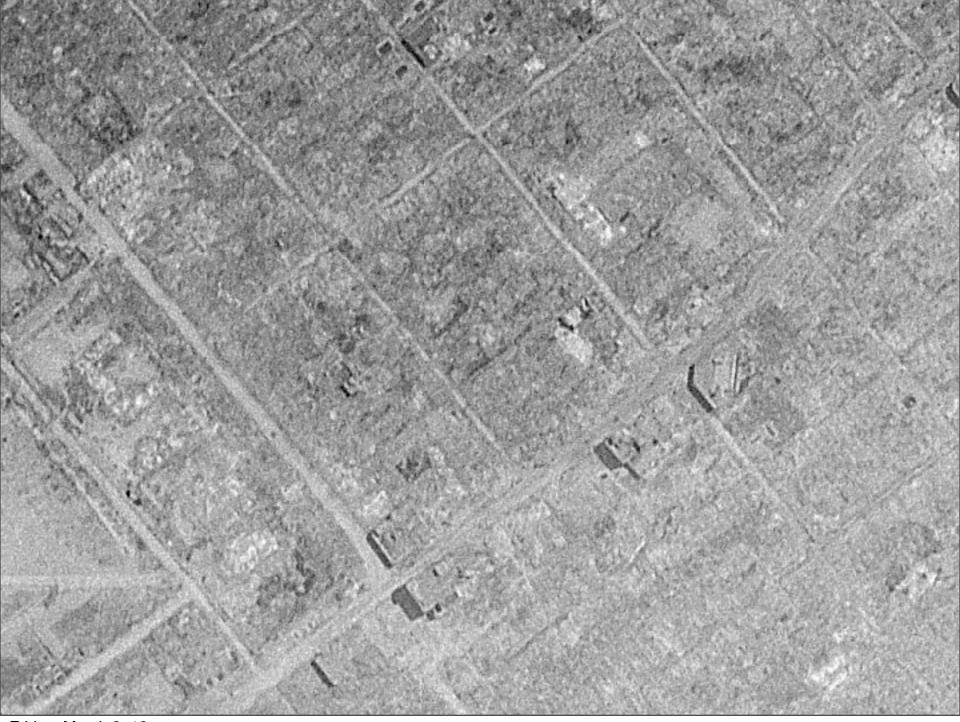
Friday, March 8, 13



Friday, March 8, 13



Friday, March 8, 13



Friday, March 8, 13

Clearly a chemical explosion occurred.

But, we know that nuclear change occurs and that is what produces the heat energy for the nuclear power plant.

Sort out nuclear change vs nuclear explosion vs chemical changes

- During Fission reactions the number and type of atoms on the left hand side of the equation is
 - A) The same as the number and type of atoms on the **right** hand side of the equation, it's just that the atoms have rearranged (Law of Conservation of Mass).
 - B) The type of atom on the right hand side is the same, it is just the number of atoms that changes

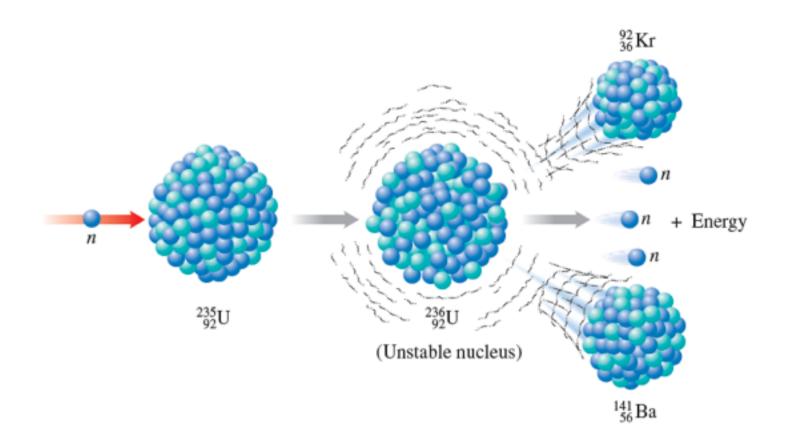


FIGURE 20.11

Upon capturing a neutron, the ²³⁵₉₂U nucleus undergoes fission to produce two lighter nuclides, free neutrons (typically three), and a large amount of energy.

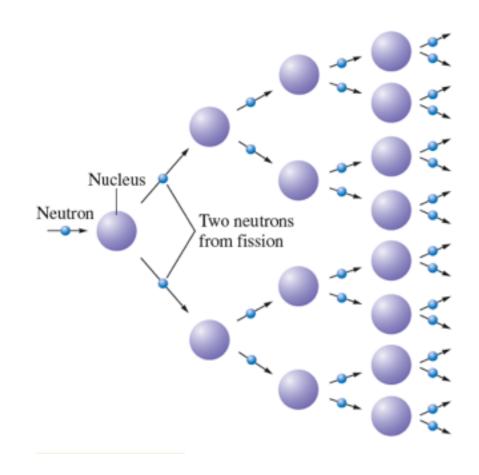


FIGURE 20.12

Representation of a fission process in which each event produces two neutrons, which can go on to split other nuclei, leading to a self-sustaining chain reaction.

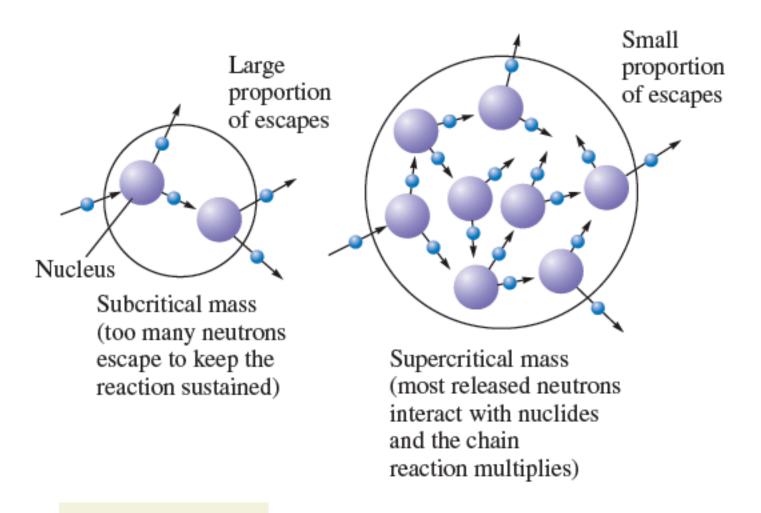


FIGURE 20.13

If the mass of fissionable material is too small, most of the neutrons escape before causing another fission event; thus the process dies out.