FINAL EXAM

signature:

Be sure and write your name on this test copy. Turn in ALL materials (exam, bubblesheet, and scratch paper) when you are finished with the exam. A periodic table and other constants are located on the bubble sheet.

Unit 1	Unit 4
$PV = nRT$ $x_{\rm A} = P_{\rm A}/P_{\rm total}$	$\Delta U = q + w \qquad \qquad H = U + PV$
P(V - nb) = nRT	$w = -P\Delta V$ $w = -\Delta nRT$
$\left(P + a\frac{n^2}{V^2}\right)(V - nb) = nRT$	$\Delta U = \Delta H - P \Delta V$
$P_{\rm total} = P_{\rm A} + P_{\rm B} + P_{\rm C} + \cdots$	$\Delta U = \Delta H - \Delta n R T$
$v_{\rm rms} = \sqrt{\frac{3RT}{M}}$ $E_{\rm k} = U = \frac{3}{2}RT$	$\Delta U = q_{\rm v} = n C \Delta T$
	$\Delta H = q_{\rm p} = n C \Delta T$
Unit 2	$q_{\rm cal} = q_{\rm water} + q_{\rm hardware}$ $q_{\rm sys} = -q_{\rm cal}$
$E = h\nu$ $c = \lambda \cdot \nu$	$\Delta S = q_{\rm rev}/T \qquad \qquad S = k \ln \Omega$
$\frac{1}{2}mv^2 = h\nu - \Phi$	$\Delta S = n C \ln \left(\frac{T_2}{T_1}\right)$
Rydberg : $\nu = \mathcal{R}\left(\frac{1}{n_1^2} - \frac{1}{n_2^2}\right)$	$\Delta H_{\rm rxn} = \Delta H_1 + \Delta H_2 + \Delta H_3 + \cdots$
$\mathcal{R} = 2.178 \times 10^{-18} \text{ J}$	$\Delta H_{\rm rxn}^{\circ} = \sum n \Delta H_{\rm f}^{\circ}({\rm prod}) - \sum n \Delta H_{\rm f}^{\circ}({\rm react})$
$\mathcal{R} = 1.097 \times 10^7 \text{ m}^{-1}$	$\Delta H_{\rm rxn} = \sum B E_{\rm breaking} - \sum B E_{\rm making}$
$\mathcal{R} = 3.29 \times 10^{15} \text{ s}^{-1}$	$\Delta G_{\rm rxn}^{\circ} = \sum n \Delta G_{\rm f}^{\circ}({\rm prod}) - \sum n \Delta G_{\rm f}^{\circ}({\rm react})$
Unit 3	$\Delta S^{\circ}_{\rm rxn} = \sum n S^{\circ}({\rm prod}) - \sum n S^{\circ}({\rm react})$
(no formulas for unit 3)	$G = H - TS$ $\Delta G = \Delta H - T\Delta S$
	$\Delta S_{\rm trans} = \Delta H_{\rm trans} / T_{\rm trans}$